



Triplex signal amplification for electrochemical DNA biosensing by coupling probe-gold nanoparticles–graphene modified electrode with enzyme functionalized carbon sphere as tracer

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ABSTRACT

An ultrasensitive electrochemical DNA biosensor was constructed by assembling probe labeled gold nanoparticles (ssDNA–AuNP) on electrochemically reduced graphene oxide (ERGO) modified electrode with thiol group tagged (GT) DNA strand (d(GT)₂₉SH) and coupling with horseradish peroxidase (HRP) functionalized carbon sphere (CNS) as tracer. The heteronanostructure formed on the biosensor surface appeared relatively good conductor for accelerating the electron transfer, while the HRP tagged CNS provided dual signal amplification for electrochemical biosensing. The triplex signal amplification strategy produced an ultrasensitive electrochemical detection of DNA down to attomolar level (5 aM) with a linear range of 5 orders of magnitude (from 1×10^{-17} M to 1×10^{-13} M), and appeared high selectivity to differentiate single-base mismatched and three-base mismatched sequences of DNA. The proposed approach provided a simple and reliable method for DNA detection with high sensitivity and specificity, indicating promising application in bioanalysis and biomedicine.

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1. Introduction

The sensitive detection of DNA sequence has emerged as a hot subject of research due to its extensive applications in molecular diagnostics, food, environment, anti-terrorism and forensic science (Dong et al., 2008; Sassolas et al., 2008; Bi et al., 2010; Dong et al., 2010a). Although various approaches have been employed for detection of low-abundant DNA, such as polymerase-chain-reaction (PCR) technique (Ju et al., 2003; Ye et al., 2003), enzymatic cycles (Munge et al., 2005; Wang et al., 2008), and nanoparticles-based signal amplification (Taton et al., 2000; Song et al., 2009), novel detection technology or signal amplification strategies are continually needed to meet the increasing demands (Wang et al., 2008). Multiple signal amplification strategy opens new opportunities for highly sensitive detection of biomolecules (Li et al., 2010). For example, two dual signal amplification strategies have been developed for detection of DNA hybridization by combining the efficient carrier-bead amplification platform with sensitive stripping voltammetric measurement (Dong et al., 2010b) and carbon nanotubes carrier amplification approach with enzymatic cycle (Gao et al., 2011).

In electrochemical biosensing, the electrocatalytic properties are strongly related to the microstructure and surface chemistry of biosensors. Recently, graphene has been used for construction of biosensors due to its unique electronic properties and large accessible surface area (Stankovich et al., 2006; Li et al., 2008; Zhou et al., 2009). A novel electrochemical immunosensor with greatly enhanced sensitivity for detection of α -fetoprotein (AFP) based on graphene modified electrode has been developed (Du et al., 2010). Biocompatible graphene not only facilitates bimolecular binding but also accelerates electron transfer, thus amplifies electrochemical detection signal (Kang et al., 2009; Wang et al., 2009a). Furthermore, the graphene modified electrode displays a superior biosensing performance in comparison to carbon nanotubes modified electrode (Alwarappan et al., 2009).

This work used electrochemically reduced graphene oxide (ERGO) modified electrode to assemble probe labeled gold nanoparticles (ssDNA–AuNP) with thiol group-DNA strand (d(GT)₂₉SH) and develop a triplex signal amplification strategy for DNA detection by coupling with streptavidin (SA)–horseradish peroxidase (HRP) functionalized carbon sphere (CNS) as a signal tag. The ERGO modified electrode was prepared by electrochemical reduction of graphene oxide (GO) coated on a glassy carbon electrode (Guo et al., 2009). The d(GT)₂₉SH was then assembled on the electrode by the π – π stacking interaction between d(GT)₂₉ and ERGO (Zheng et al., 2003) for anchoring ssDNA–AuNPs to

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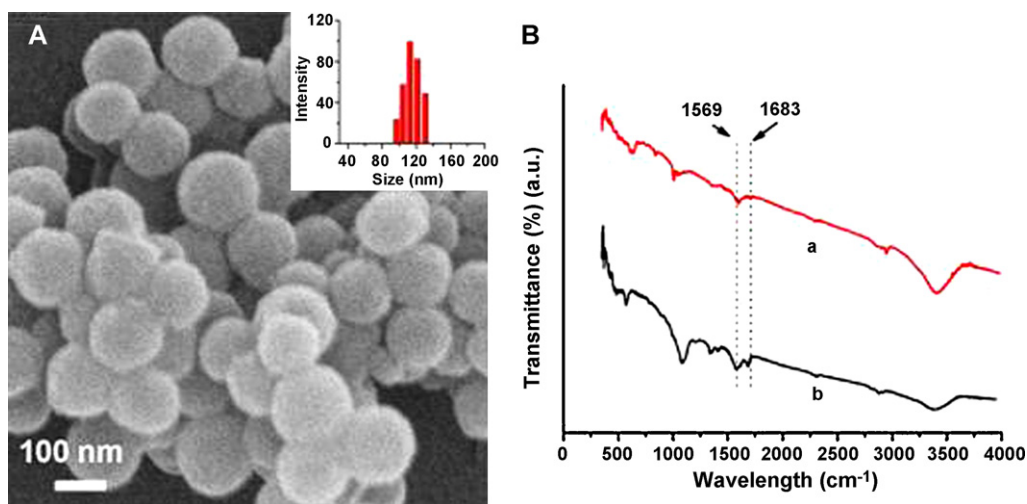


Fig. 2. (A) SEM of as-prepared CNS and (B) FT-IR spectra of CNS (a) and CNS-SA/HRP (b). Inset in A: DLS of prepared CNS.

3. Results and discussion

3.1. Characterization of CNS and SA-HRP tagged CNS

The CNS was obtained by a “green” synthetic approach, and functionalized with hydrophilic carboxylate groups by sonication

in 6:1 concentrated H_2SO_4/HNO_3 for 1 h. The carboxylated CNS showed a homogeneous and smooth surface and good dispersion with a narrow size distribution (Fig. 2A). Their mean size was about 120 nm. The DLS further confirmed the narrow size distribution of the obtained CNS (inset in Fig. 2A). The hydrophilic surface could be easily functionalized with SA-HRP by covalent linking. In

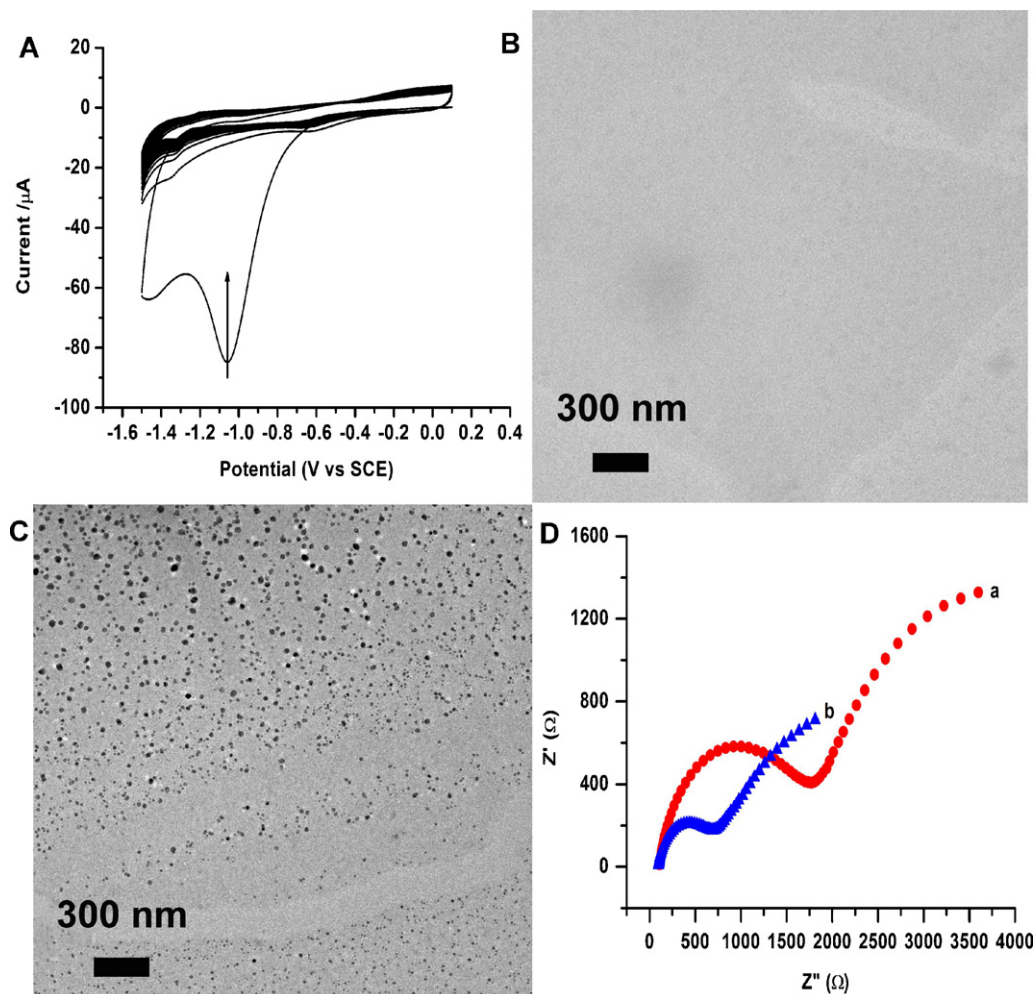


Fig. 3. (A) CV of GO modified GCE in nitrogen-saturated PBS (pH 5.2) at 50 mV/s, (B) TEM image of ERGO, (C) TEM image of ssDNA-AuNPs loaded ERGO, and (D) electrochemical impedance spectra of (a) GO and (b) ssDNA-AuNPs assembled ERGO modified GCEs in 0.1 M KCl containing 5 mM $K_3[Fe(CN)_6]/K_4[Fe(CN)_6]$.

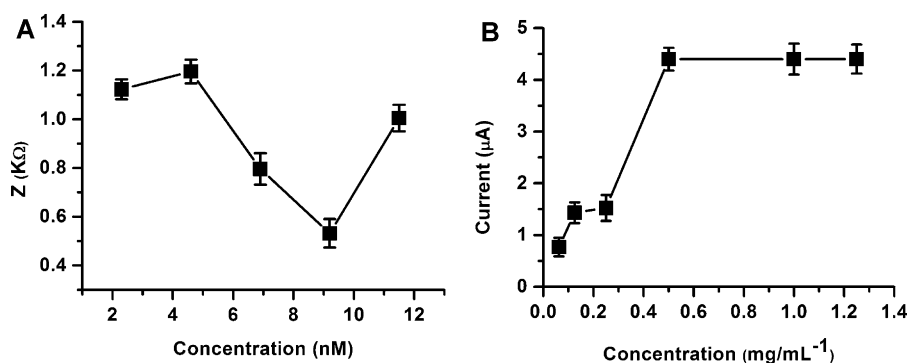


Fig. 4. (A) Electrochemical impedance at different concentrations of ssDNA–AuNPs and (B) plot of peak current vs HRP–CNS concentration.

comparison with the CNS (Fig. 2B, curve a), the Fourier transform infrared spectrum of the SA–HRP tagged CNS displayed obvious absorption peaks corresponding to the amide bands I (1683 cm^{-1}) and II (1569 cm^{-1}) of SA–HRP (Fig. 2B, curve b), respectively, indicating that the SA–HRP molecules were successfully loaded on the carboxylated CNS.

3.2. Characterization of electrode assembly process

The cyclic voltammogram (CV) of GO modified GCE showed a large reduction peak around -1.1 V , which was related to the reduction of the surface oxygen groups. The reduction current decreased considerably at the second cycle and disappeared after several potential scans (Fig. 3A). This phenomenon indicated that the reduction of surface-oxygenated groups on GO was quick and irreversible (Guo et al., 2009; Wang et al., 2009b; Shao et al., 2010). Transmission electron microscopic images showed few-layer flexible wrinkled sheet of ERGO (Fig. 3B). Benefiting from the strong π – π stacking between $d(\text{GT})_{29}\text{SH}$ and ERGO, the ssDNA–AuNPs could be efficiently loaded on the surface of ERGO via Au–S bond (Fig. 3C). Furthermore, the colloidal AuNPs were distributed on the whole surface of ERGO sheets. It is worthy to mention that the direct assembly of DNA on heteronanostructure did not alter the structures and properties of both AuNPs and ERGO, including their high conductivity for accelerating the electron transfer. As shown in Fig. 3D, the GO modified GCE showed an electron-transfer resistance of about $2400\ \Omega$, while ssDNA–AuNPs assembled ERGO modified GCE displayed a resistance of only about $800\ \Omega$ in spite of the presence of high content of ssDNA with negative charge of phospholipids, which would block the electron transfer of negatively charged $\text{Fe}(\text{CN})_6^{3-/4-}$. The low resistance indicated the good conductivity of the ssDNA–AuNPs assembled ERGO.

3.3. Optimization of biosensing condition

The amount of ssDNA–AuNPs attached on the ERGO sheet and the concentration of the tracer affected intensely the electrochemical performance of the proposed sensor for detection DNA target (Fig. 4). The electron-transfer resistance decreased along with the increase of the Au–ssDNA concentration with the lowest value observed at 9.2 nM (Fig. 4A). The further increase of Au–ssDNA caused the increasing electron-transfer resistance, which might result from the citrate moieties on AuNPs (Gao et al., 2011). The electrochemical response was intensified in a tracer concentration-dependent manner, and tended to a stable value at 0.5 mg/mL (Fig. 4B). Thus, 9.2 nM ssDNA–AuNPs and 0.5 mg/mL SA–HRP–CNSs were used in all subsequent works, respectively.

3.4. Signal amplification

After the proposed biosensor hybridized with the biotinylated target and incubated with SA–HRP tagged CNS solution for 30 min, respectively, it exhibited a stable DPV peak at -0.564 V in PBS containing *o*-phenylenediamine and hydrogen peroxide, which corresponded to the reduction of 2,2'-diaminoazobenzene, the enzymatic product (Cheng et al., 2008). The tracer made each target hybridization event relate to numerous enzyme reaction, thus greatly amplify the detection signal. As a comparison, SA–HRP was used as the tracer instead of SA–HRP–CNS. As shown in Fig. 5, the signal originated from SA–HRP–CNS tracer (curve a) was 3.6-fold larger than that from SA–HRP (curve b) at 0.1 fM target DNA, and 6.8-fold larger than the background (curve c), which was detected in absence of target and came from the nonspecific adsorption of SA–HRP–CNS. This result demonstrated that the SA–HRP–CNS tracer greatly amplified the detection signal.

3.5. Voltammetric analysis of target DNA

Under optimal conditions, the DPV peak current increased with the increasing concentration of target DNA (Fig. 6A). The plot of the response vs the logarithm of target concentration showed a linear relationship in the detected range from $1 \times 10^{-17}\text{ M}$ to $1 \times 10^{-13}\text{ M}$ (inset in Fig. 6A). At the current signal of 3 times standard deviation (SD) measured at free of target DNA, the limit of detection (LOD) was calculated to be 5 aM . Compared to other enzyme-based electrochemical DNA sensors (Munge et al., 2005; Wang et al., 2008), the proposed biosensor showed a much lower limit of detection. The limit of detection was also competitive with other highly

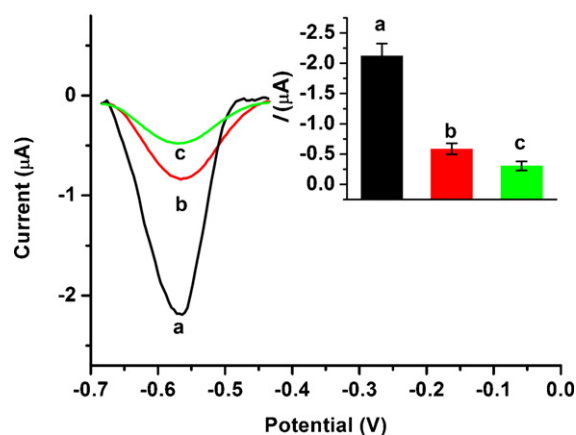


Fig. 5. DPV curves of (a) 0.1 fM target with SA–HRP–CNS tracer, (b) 0.1 fM target with SA–HRP tracer and (c) blank with SA–HRP–CNS tracer. Inset: the corresponding peak currents.

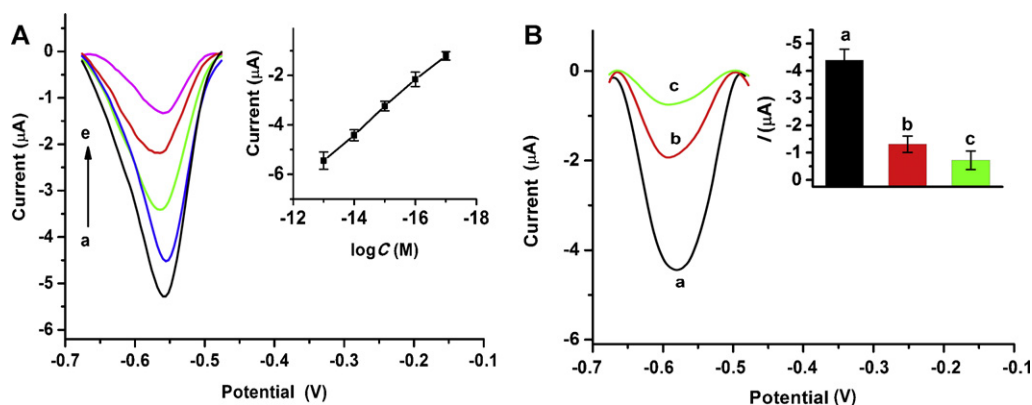


Fig. 6. (A) DPV curves at target DNA concentrations of 0.1, 0.01 pmol/L, 0.1, 0.01 fmol/L from (a) to (e) and (B) DPV curves at 0.01 pM complementary sequence (a), single-base mismatched sequence (b) and three-base mismatched sequence (c). Inset in A: linear relationship between peak current and logarithm of target concentration ($n = 3$). Inset in B: the corresponding current values.

sensitive detection approaches such as molecular biological technique (Saghatelian et al., 2003; Tan et al., 2005; Mahmoudian et al., 2008), and nanoparticle-based signal amplification strategy (Liu et al., 2008; Pinijsuwan et al., 2008; Rijiravanich et al., 2008). The high sensitivity emphasized the importance of the proposed triplex signal amplification strategy to the improved analytical performance of the biosensor.

The specificity of the proposed DNA biosensor was investigated by exposing it to three kinds of DNA sequences, including perfect complementary target, single-base mismatched oligonucleotide and three-base of mismatched oligonucleotide at the same concentration (0.01 pM). The biosensor exhibited good performance to discriminate perfect complementary target and the base mismatched targets (Fig. 6B). The perfect complementary target showed a signal of 3.4-fold and 6.6-fold of those single-base mismatched oligonucleotide and three-base mismatched oligonucleotide, respectively, indicating good selectivity and great potential for single nucleotide polymorphism analysis.

4. Conclusions

This work has demonstrated a highly sensitive electrochemical DNA detection platform based on a triplex signal amplification strategy using probe labeled gold nanoparticles–graphene modified electrode and enzyme functionalized carbon sphere as tracer. The DNA mediated noncovalent assembly of AuNPs–graphene heteronanostructure does not alter the structures and properties of AuNPs and ERGO, which makes the ssDNA–AuNPs–ERGO maintain relatively good conductivity. The SA–HRP–CNS tracer enables the hybridization event relating to numerous enzyme reactions, providing dual signal amplification for ultrasensitive detection of DNA. This novel triplex signal amplification strategy can detect target DNA down to attomolar level with high selectivity to differentiate single-base mismatched and three-base mismatched sequences of DNA. The proposed approach would be attractive for genetic target analysis in bioanalytical and clinic biomedical application.

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References

- Alwarappan, S., Erdem, A., Liu, C., Li, C.Z., 2009. *J. Phys. Chem. C* 113, 8853–8857.
- Bi, S., Zhang, J.L., Zhang, S.S., 2010. *Chem. Commun.* 46, 5509–5511.
- Cheng, W., Ding, L., Lei, J.P., Ding, S.J., Ju, H.X., 2008. *Anal. Chem.* 80, 3867–3872.
- Dong, H.F., Gao, W.C., Yan, F., Ji, H.X., Ju, H.X., 2010a. *Anal. Chem.* 82, 5511–5517.
- Dong, H.F., Yan, F., Ji, H.X., Wong, D.K.Y., Ju, H.X., 2010b. *Adv. Funct. Mater.* 20, 1173–1179.
- Dong, X.C., Lau, C.M., Lohani, A., Mhaisalkar, S.G., Kasim, J., Shen, Z.X., Ho, X.N., Rogers, J.A., Li, L.J., 2008. *Adv. Mater.* 20, 2389–2393.
- Du, D., Zou, Z.X., Shin, Y., Wang, J., Wu, H., Engelhard, M.H., Liu, J., Aksay, I.A., Lin, Y.H., 2010. *Anal. Chem.* 82, 2989–2995.
- Gao, W.C., Dong, H.F., Lei, J.P., Ji, H.X., Ju, H.X., 2011. *Chem. Commun.* 47, 5220–5222.
- Guo, H.L., Wang, X.F., Qian, Q.Y., Wang, F.B., Xia, X.H., 2009. *ACS Nano* 3, 2653–2659.
- Ju, H.X., Ye, Y.K., Zhao, J.H., Zhu, Y.L., 2003. *Anal. Biochem.* 313, 255–261.
- Kang, X.H., Wang, J., Kang, X.H., Wu, H., Aksay, I.A., Liu, J., Lin, Y.H., 2009. *Biosens. Bioelectron.* 25, 901–905.
- Li, J.S., Deng, T., Chu, X., Yang, R.H., Jiang, J.H., Shen, G.L., Yu, R.Q., 2010. *Anal. Chem.* 82, 2811–2816.
- Li, X.L., Wang, X.R., Zhang, L., Lee, S.W., Dai, H.J., 2008. *Science* 319, 1229–1232.
- Liu, J.W., Lu, Y., 2006. *Nat. Protoc.* 1, 246–252.
- Liu, J.B., Li, Y.L., Li, Y.M., Li, J.H., Deng, Z.X., 2010. *J. Mater. Chem.* 20, 900–906.
- Liu, G., Wan, Y., Gau, V., Zhang, J., Wang, L.H., Song, S.P., Fan, C.H., 2008. *J. Am. Chem. Soc.* 130, 6820–6825.
- Munge, B., Liu, G.D., Collins, G., Wang, J., 2005. *Anal. Chem.* 77, 4662–4666.
- Mahmoudian, L., Kaji, N., Tokeshi, M., Nilsson, M., Baba, Y., 2008. *Anal. Chem.* 80, 2483–2490.
- Pinijsuwan, S., Rijiravanich, P., Somasundrum, M., Surareungchai, W., 2008. *Anal. Chem.* 80, 6779–6784.
- Rijiravanich, P., Somasundrum, M., Surareungchai, W., 2008. *Anal. Chem.* 80, 3904–3909.
- Saghatelian, A., Guckian, K.M., Thayer, D.A., Ghadiri, M.R., 2003. *J. Am. Chem. Soc.* 125, 344–345.
- Sassolas, A., Leca-Bouvier, B.D., Blum, L.J., 2008. *Chem. Rev.* 108, 109–139.
- Shao, Y.Y., Wang, J., Engelhard, M., Wang, C.M., Lin, Y.H., 2010. *J. Mater. Chem.* 20, 743–748.
- Shin, Y.S., Wang, L.Q., Bae, I.T., Arey, B.W., Exarhos, G.J., 2008. *J. Phys. Chem. C* 112, 14236–14240.
- Song, S.P., Liang, Z.Q., Zhang, J., Wang, L.H., Li, G.X., Fan, C.H., 2009. *Angew. Chem. Int. Ed.* 48, 8670–8667.
- Sun, X., Li, Y., 2004. *Angew. Chem. Int. Ed.* 43, 597–601.
- Stankovich, S., Dikin, D.A., Dommett, G.H.B., Kohlhaas, K.M., Zimney, E.J., Stach, E.A., Piner, R.D., Nguyen, S.T., Ruoff, R.S., 2006. *Nature* 442, 282–286.
- Tan, E., Nguyen, J., Wong, D., Zhang, Y., Erwin, B., Ness, L.K.V., Baker, S.M., Galas, D.J., Niemi, A., 2005. *Anal. Chem.* 77, 7984–7992.
- Taton, T.A., Mirkin, C.A., Letsinger, R.L., 2000. *Science* 289, 1757–1760.
- Wang, Y., Lu, J., Tang, L.H., Chang, H.X., Li, J.H., 2009a. *Anal. Chem.* 81, 9710–9715.
- Wang, J.X., Zhu, X., Tu, Q.Y., Guo, Q., Zarui, C.S., Momand, J., Sun, X.Z., Zhou, F.M., 2008. *Anal. Chem.* 80, 769–774.
- Wang, Z.J., Zhou, X.Z., Zhang, J., Boey, F., Zhang, H., 2009b. *J. Phys. Chem. C* 113, 14071–14075.
- Ye, Y.K., Zhao, J.H., Yan, F., Zhu, Y.L., Ju, H.X., 2003. *Biosens. Bioelectron.* 18, 1501–1508.
- Zheng, M., Jagota, A., Semke, E.D., Diner, B.A., Mclean, R.S., Lustig, S.R., Richardson, R.E., Tassi, N.G., 2003. *Nat. Mater.* 3, 8–342.
- Zhou, M., Zhai, Y.M., Dong, S.J., 2009. *Anal. Chem.* 81, 5603–5613.